The new features in the trunk some above ground and some belowground

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ORCHIDEE-CAN

(known as ORCHIDEE-DOFOCO on svn)

ORCHIDEE-CN

N-version of ORCHIDEE updated with the trunk, June 2017



ORCHIDEE-CN-CAN

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Pipe model theory

- Recognize how stomata is hydrological connected to the roots and the need to invest carbon in building roots and stem
- Allometric relationships, . leaf to sapwood area ratio, relationship between diameter and height

Water stress

Hydraulic architecture .



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Soil Y



Courtesy of S. Luyssaert

- Diameter classes and age classes are introduced
- Number of PFTs depend on number of age classes
- Each PFT has x numbers of diameter class
- Each diameter class has x number of trees depending on basal area - self-thinning rule





- The trees are horizontally distributed following a Poisson distribution
- The structured canopy allows for calculations of light penetration within the canopy.
- Statistic approach to reduce memory allocation



ECOSYSTEM DYNAMIC



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ECOSYSTEM DYNAMIC



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ECOSYSTEM DYNAMIC



The soil biogeochemistry in ORCHIDEE

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- Split between stomate_litter.f90 and stomate_soilcarbon.f90
- Run at ½ hourly time-step whereas stomate runs at daily time-step.

 Moisture and temperature function calculated in stomate_litter.f90

$$\tau = Q_{10}^{(T-Topt)/10}$$

$$\theta = Max(0.25, Min(1, M))$$

 $M = -1.1 * SM^2 + 2.4 * SM - 0.29$

- Input from plants
 through *bm_to_litter* and *turnover*
- Split between above and below ground
- Split into two pools:
 metabolic/structural
 depending on lignin and
 N content of the litter.



- Inputs from litter
 decomposition in
 soilcarbon_input
- Distributed into the active and slow pools control by the lignin content.



 Decomposition Structural Litter Litter following 1st order Metabolic kinetics. Litter CO_2 $\frac{\partial SOC}{dt} = I - k \times SOC \times \theta \times \tau$ CO_2 Active SOM ∂t CO_2 CO_2 A fraction of C CO_2 Slow SOM CO₂ decomposed is respired CO_2 CO_2 the 1-resp is distributed Passive SOM in the other pools.



HOW THE NITRIFICATION/DENITRIFICATION PROCESSES ARE REPRESENTED



Peng and Zhu (2006)

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HOW THE NITRIFICATION/DENITRIFICATION PROCESSES ARE REPRESENTED

- Key point -> N outputs fluxes & GHG production
- DNDC is an old model based on Li et al. 1992.
- Design to represent denitrification <u>and</u> decomposition.

 In ORCHIDEE, only the N-related aspects are used but in a simplified way.





Soil temperature & moisture

from submodel of thermal-

hydraulic flow







IMPROVED DNDC

The PnET-N-DNDC Model



IMPROVED DNDC



IMPROVED DNDC



THE ANAEROBIC BALLOON CONCEPT



THE ANAEROBIC BALLOON CONCEPT

 Table 2. Functions and Parameters for O2 Diffusion and Volumetric Fraction of Anaerobic Microsites (ANVF)

Equation No.	Function	Equation
1	oxygen diffusion coefficient in soil	$D_{s[L]} = D_{air} afps_{[L]}^{3.33} / afps_{max[L]}^{2.0};$
2	oxygen diffusion rate affected by frost	
		$ \begin{array}{ll} D_{s[L]} = D_{s[L]} & F_{frost}; & 0 < D_{s[L]} < 1 \\ & \text{if } T > 0 \ ^{\circ}\text{C}, & F_{frost} = 1.2; \\ & \text{if } T <= 0 \ ^{\circ}\text{C}, & F_{frost} = 0.8; \end{array} $
3	oxygen partial pressure	$d(pO_{2[L]})/dt = (d(D_{s[L]} d(pO2_{[L]})/dz)/dz - R)/afps;$
4	Volumetric fraction of anaerobic microsites	$anvf_{[L]} = a (1-(b pO_{2[L]}/pO_{2air}));$

a, b, constant coefficients; afps, air-filled porosity; afps_{max}, porosity; anvf, volumetric fraction of anaerobic microsites; D_{air} , oxygen diffusion rate in the air, 0.07236 m²/h [*Beisecker*, 1994]; D_s , oxygen diffusion coefficient in soil; F_frost, frost factor; L, layer number; pO₂, oxygen partial pressure; R, oxygen consumption rate (kg C ha⁻¹h⁻¹); t, time (h); z, soil depth (m).

THE EQUATIONS

Table 3. Function	and Parameters	for Nitrification
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Equation No.	Function	Equation
1	relative growth rate of nitrifiers	$\mu_{a} = \mu_{MAX} ((DOC] / (1 + [DOC]) + F_{m} / (1 + F_{m}));$
2	relative death rate of nitrifiers	$\mu_{\rm d} = a_{\rm MAX} B_{\rm n} / (5 + [\rm DOC]) / (1 + F_{\rm m});$
3	net increase in nitrifiers biomass	$\mu_{\rm b} = (\mu_{\rm g} - \mu_{\rm d}) \mathbf{B}_{\rm n} \mathbf{F}_{\rm t} \mathbf{F}_{\rm m};$
4	nitrification rate	$\mathbf{R}_{n} = \mathbf{R}_{max} [\mathbf{NH4}] \mathbf{B}_{n} \mathbf{pH};$
5	temperature factor	$F_t = ((60-T)/(25.78)^{3.503} e^{(3.503 (T-34.22)/(25.78))};$
6	moisture factor	if wfps > 0.05 $F_m = 1.01 - 0.21$ wfps; if wfps <= 0.05 $F_m = 0$;
7	NO production from nitrification	$NO = .0025 R_n F_t;$
8	N ₂ O production from nitrification	$N_2O = 0.0006 R_n F_t$ wfps;

 a_{MAX} , maximum death rate for nitrifiers (1.44 1/d [from *Blagodatsky and Richter*, 1998]); B_n , biomass of nitrifiers (kg C/ha); [DOC], concentration of dissolved organic C (kg C/ha); F_m , moisture factor; F_v , temperature factor; [NH4], concentration of ammonium (kg N/ha); NO, NO production from nitrification; N₂O, N₂O production from nitrification [*Ingwersen et al.*, 1999]; pH, soil pH; R_n , nitrification rate; R_{max} , maximum nitrification rate (1/h); T, soil temperature (°C); wfps, water-filled porosity; μ_{MAX} , maximum growth rate for nitrifiers (4.87 1/d [from *Blagodatsky and Richter*, 1998]); μ_b , net increase in nitrifiers biomass; μ_d , relative death rate of nitrifiers; μ_g , relative growth rate of nitrifiers.

THE EQUATIONS

Equation No.	Function	Equation
1	relative growth rate of Nox denitrifiers	$\mu_{NOx} \doteq \mu_{NOx(max)} \text{ [DOC]/(Kc+[DOC]) [No_x]/(Kn+[NO_x]);}$
2	relative growth rate of total denitrifiers	$\mu_{g} = F_{t} (\mu_{NO3} F_{PH1} + \mu_{NO2} F_{PH2} + \mu_{NO} F_{PH2} + \mu_{N20} F_{PH3};$ $F_{t} = 2^{((T-22.5)/10)};$ $F_{PH1} = 1 - 1 / (1 + e^{(pH-4.25/0.5)});$ $F_{PH2} = 1 - 1 / (1 + e^{(pH-4.25/1.0)});$ $F_{PH3} = 1 - 1 / (1 + e^{(pH-6.25/1.5)});$
3	denitrifier growth rate, death rate, and consumption rate of soluble carbon	$R_{g} = \mu_{g} B_{d};$ $R_{d} = M_{c} Y_{c} B_{d};$ $R_{C} = (\mu_{g} / Y_{c} + M_{c}) B_{d};$
4	consumption rates of N oxides	$\mathbf{R}_{NOx} = (\mu_{NOx}/Y_{NOx} + M_{NOx} [No_x]/[N]) \mathbf{B}_d;$
5	nitrogen assimiliation rate	$q_N = R_g / CN;$
6	gas diffusion factor	$v = D_{max}$ afps (1 - anvf) $F_{clay} 2^{T/20}$; $F_{rac} = 0.13 - 0.079$ clay:

Table 4. Functions and Parameters for Denitrification

afps, air-filled porosity; anyf, volumetric fraction of anaerobic microsites; B_{dy} denitrifier biomass (kg C/m³); clay, clay fraction in the soil; CN, C/N ratio in denitrifiers (3.45 [Van Verseveld and Stouthamer 1978]); D_c, consumption rate of soluble carbon by denitrifiers (kg C m⁻³h⁻¹); D_{max}, maximum diffusion rate in air (m^2/h) ; D_{NOS} , consumption rate of N oxides by denitrifiers (kg C m³h⁻¹); [DOC], so!uble C concentration (kg C/m³); F_{elav}, clay factor; F_v, temperature factor; F_{PH1}, pH factors for NO₃ denitrifiers; F_{PH2}, pH factors for NO₂ and NO denitrifiers; F_{Pita}, pH factors for N₂O denitrifiers; Kc, half-saturation value of soluble carbon (0.017 kg C/m³ [Shan and Coulman, 1978]); Kn, half-saturation value of N oxides (0.083 kg N/m³ [Shan and Coulman, 1978]); M_c, maintenance coefficient on carbon (0.0076 kg N kg⁻¹h⁻¹ [Van Verseveld et al., 1977]); [N], concentration of all NO_x (kg N/m³); [No_x], concentration of for NO₃, NO₂, NO and N₂O (kg N/m³); pH, soil pH; q_n , nitrogen assimilation rate (kg N ha⁻¹h⁻¹); T, soil temperature (°C); v, gas diffusion factor (%); Y_n , maximum growth rate of denitrifiers on soluble carbon (0.503 kg C/kg C [Van Verseveld et al., 1977]); M_{NOx}, maintainance coefficient on N oxides (0.09, 0.035 and 0.079 kg N/kg/ for NO₃⁻, NO₂⁻ (+NO*) and N₂O, respectively, based on Van Verseveld et al. [1977]); R_d , denitrifier death rate; R_e , denitrifier growth rate; Y_{NOx} , maximum growth rate on N oxides (0.401, 0.428 and 0.151 kg C/kg N for NO₂⁻, NO₂⁻ (+NO*) and N₂O, respectively, based on Van Verseveld et al. [1977]); μ_a , relative growth rate of total denitrifiers (1/h); μ_{NO3} , μ_{NO2} . μ_{NO}, μ_{N2O}, relative growth rate of NO₃⁻, NO₂⁻, NO-, and N₂O denitrifiers; μNOx, relative growth rate of NO_x denitrifiers (1/h); µ_{NOx(max)}, maximum growth rates (0.67 1/h for NO₃, NO₂ denitrifiers, and 0.34 1/h for NO and N₂O denitrifiers, based on *Hartel and Alexander* [1987]). The parameters are shared by NO₂ and NO due to the lack of data for NO.

WHAT IS DONE IN ORCHIDEE

- Implemented by Zaehle et al., (2010)
- At that time, no DOC and no soil C discretization.
- Active C pools used instead of DOC.
- Gas diffusion is calculated using a fixed soil depth value of 20cm.
- Several parameters tuned.
- With Nicolas, we decided to let the original parameter values.

A NEW SOIL SCHEME FROM THE MICT BRANCH

- For CMIP6, some LSMs will have permafrost C, explicit N cycle and perhaps both
- All the necessary piece of code exist within the « orchidee environment »
- Opportunity to benefit from the huge effort done in the MICT branch by many colleagues (Dan Zhu, Philippe Ciais, Matthieu Guimberteau, Charlie Koven, …)

Only an option in the trunk controlled by OK_SOIL_CARBON_DISCRETIZATION

- Several options are available in MICT but not in the trunk (fire, grassland management, permafrost C)
- Focus on permafrost C
- Not only adding « frozen C »
- Soil C is discretized
- Diffusion is added (including bioturbation and cryoturbation)
- Temperature effect on SOC mineralization
- When frozen, nroot is set to -> impact on water stress and on transpiration.
- Optional
 - Zimov effect
 - Insolation effect (thermal conductivity affected by SOC)



Guimberteau et al 2018



Guimberteau et al 2018



Guimberteau et al 2018

WHAT IS NEW

- Soil organic N is also discretized
- Mineral N is not
- No effect on N plant uptake

- Representation of the soil C/N profile
- Lateral outputs of C (DOC, Erosion)
- Representation of Priming effect.
- Carbon isotopes (¹⁴C and ¹³C)
- Peatlands

THANK YOU FOR YOUR ATTENTION!

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